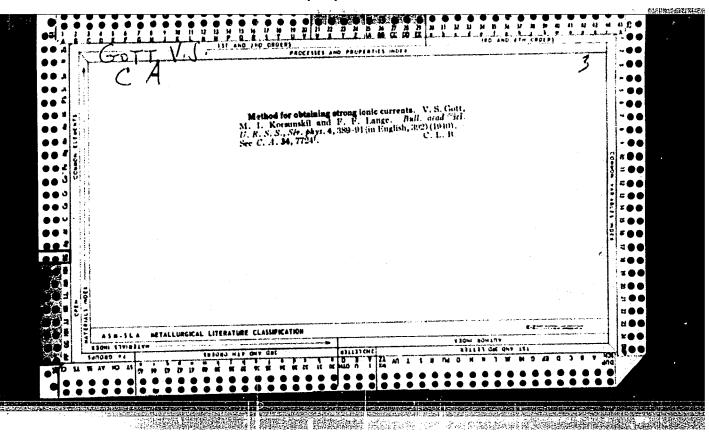
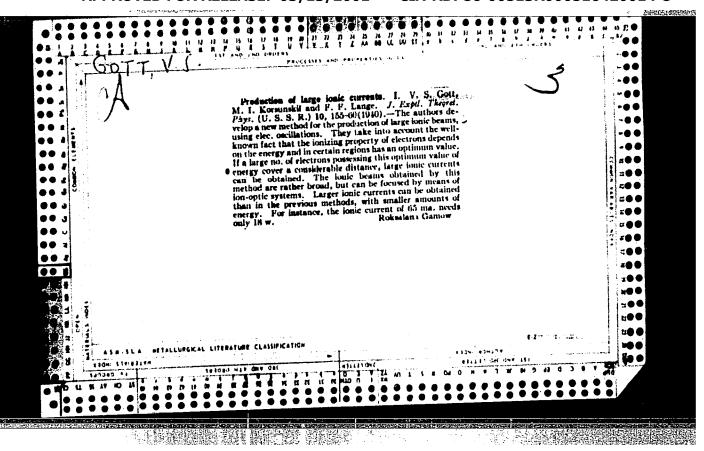
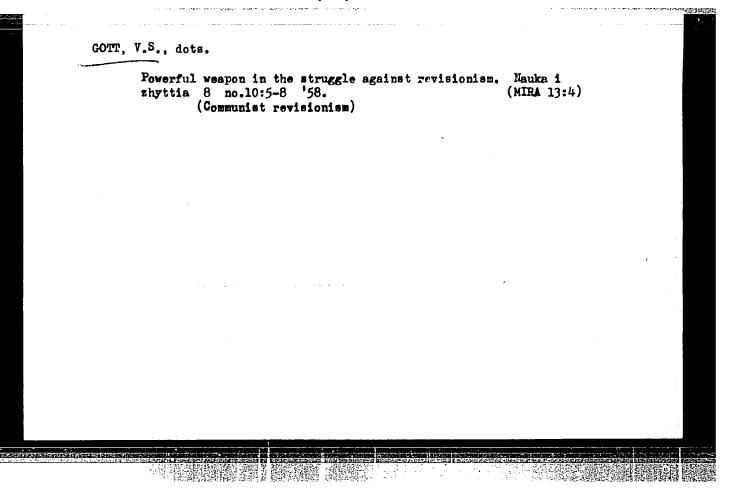
GOTT, H.H.; MASTERS, E.F.O.; SHIPP, G.C.; KOCZOGH, Akosne [translator]

Reactors of atomic power plants. Atom taj 2 no.1:67-88 Ja '59.







Matter and motion. Nauka i shyttia 9 no.4:15-17 Ap '59.

(Matter) (Notion)

PYASKOVSKIY, Boleslav Vital'yevich [Piaskovs'kyi, B.V.]; GOTT, V.S.[Hott. V.S.], kand.fiziko-matem.nauk, glavnyy red.

[Struggle of meterialism against idealism in present-day mathematics] Borot'ba materializmu z idealizmom v suchasnii matematytai. Kytv, 1960. p. 39. (Tovarystvo dlia poshyrennia politychnykh i naukovykh znan' Ukrains'koi RSR. Ser.5, no.1).

(MIRA 13:6)

(Mathematics--Philosophy)

	GOTT Yu. 38. a	nd IOFFE, N. S.	. and TÖLKOVS	KY, V. G.		
	"Some ne	w results on th	e confinemen	t of magnetic	traps"	
	Report presen Nuclear Fusio	ted at the Conf n Salzburg, Aus	erence on Platria, 4-8 Se	asma Physics a p 61	nd ontrolled	
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27168 S/057/61/031/009/008/019 B104/B102

24.6710

AUTHORS:

Gott, Yu. V., and Tel'kovskiy, V. G.

TITLE:

Determination of the ionic energy in a high-temperature

plasma by means of thin foils

PERIODICAL: Zhurnal tekhnicheskoy fiziki, v. 31, no. 9, 1961, 1061-1065

TEXT: The method of measuring the ionic energies and the energies of neutral particles by means of thin foils is based on the fact that after the passage of a ray through several atomic layers, the fractions of charged and neutral particles within the foil do not depend on their original charge. The main difficulty of this method lies in the production of thin homogeneous foils. The authors produced films by sputtering silver at $10^{-5}-10^{-6}$ mm Hg on an aluminum foil (2-3 μ thick). The silver was evaporated. The temperature of the Λ l base layer exerted a great effect on the quality of foils. In the tests described the Λ l base layer was kept at nitrogen temperature. After production of the silver foil, the aluminum was removed with caustic soda. Then, the silver foil was taken out from the solution with a very fine copper net.

Card 1/3

27168 S/057/61/031/009/008/019 B104/B102

Determination of the ionic energy in ...

The resulting foils had thicknesses of 100-2000 A, and a size of 10-20 cm2. Their inhomogeneity was less than 4 %. After the ion beam has passed through the foil with an energy below 20 kev it contained many neutral particles. Behind the foil, an insulated collector was installed in front of which a grid was arranged with a potential of +40 v against the collector. Thus, both the ions and the neutral particles (due to secondary electron emission) could be recorded. Since every foil is porous, a certain current always reaches the collector at any velocity of ions. The current to the collector increases considerably from the ionic energy at which the ions begin to pass through the foil. This energy is called threshold energy. Measurements with hydrogen, deuterium, and helium ions showed this threshold energy to be proportional to the foil thickness up to a foil thickness of about 1300 A. At a foil thickness of 100 %, the energy of hydrogen and deuterium ions can be measured from 500 ev, that of helium ions from 2000 ev. The effect of ion scattering was eliminated by use of a semispherical collector. Thus, it was shown that the current to the collector may be well described by

Card 2/3

27168 S/057/61/031/009/008/019 B104/B102

Determination of the ionic energy in...

 $^{\rm I}$ coll = α (E-E_{thresh}) $_{\rm I_0}$; α = A/d $^{\rm n}$. d is the thickness of the foil; A is a constant mainly dependent on the coefficient of secondary electron emission; n = 0.85 for H $^+$, 0.42 for D $^+$, and 0.2 for He $^+$. The following is obtained for the determination of the integral spectrum:

$$\int_{E_{\text{thresh}}}^{E_{\text{max}}} I(E)dE = \frac{1}{\alpha^2} I_{\text{coll}} \frac{d\alpha}{dE_{\text{thresh}}} - \frac{1}{\alpha} \frac{dI_{\text{coll}}}{dE_{\text{thresh}}}$$

The authors thank Academician L. A. Artsimovich for valuable advice. There are 3 figures, 1 table, and 9 references: 7 Soviet and 2 non-Soviet. The reference to the English-language publication reads as follows: I. A. Phillips, Phys. Rev., 97, 2, 404, 1955.

SUBMITTED: October 17, 1960

Card 3/3

42728

S/109/62/007/011/007/012 D295/D308

AUTHORS:

Gott, Yu.V. and Tel'kovskiy, V.G.

TITLE:

Deceleration of light ions in thin

metal foils

PERIODICAL:

Radiotekhnika i elektronika, v. 7,

no. 11, 1962, 1956 - 1961

TEXT:

The passage of ions with less than 20keV energy through metal foils has been little studied. The small ion velocity makes both Bohr's and Fermi-Teller's formulas inapplicable. In the described experimental investigation a hydrogen, deuterium or helium ion beam from a high-frequency source with 1.5 - 15 keV energy is deviated by 90° by a sectorial magnetic field with double focusing and passes through 100-500 Å thick foils of Ag, Ti and Ge. The output-beam energy distribution is measured by means of an electrostatic analyzer. The specific energy losses (-dE/dx, where E is the ion energy and x is the transverse coordinate of the foil) obtained for several foils

Card 1/2

Deceleration of light ions ...

S/109/62/007/011/007/012 D295/D308

of various thickness are plotted as a function of the mean beam velocity and exhibit a linear relationship for velocities greater than 108 cm/sec. Departures from linearity below this velocity are interpreted as due to a reduction of the effective charge owing to electron capture. The linearity coefficients are independent of the ion mass and of the properties of the atoms of the target and are determined mainly by the inter-atomic distances of the crystal lattice of the target. Calculations based on a simplified model for the loss mechanism, in which the target is assimilated to a very dense plasma, are in good agreement with experiment and suggest that, for the velocity interval investigated, energy loss is due mainly to collisons with free electrons. The results of similar experiments with molecular ions (Ht and Ht) confirm this view.

These ions are dissociated at a small distance (10 - 20 Å) from the target input surface and the energy-loss calculations are to be carried out for the dissociation products separately. The most important English-language reference is: J.R. Young, J.Appl. Phys., v. 27, 1956, 1. There are 6 figures and 1 table. SUBMITTED: March 19, 1962 ' Card 2/2

GOTT, Yu.V.; TEL'KOVSKIY, V.G.

Determining the lifetime of a high-temprature plasma on the basis of the duration of neutron emission. Zhur. eksp. i teor. fiz. 43 no.3; 831-834 *62. (MIRA 15:10)

(Plasma (Ionized gases)) (Neutrons)

L 23827-65 EWT(1)/EWG(k)/EPA(sp)-2/EPA(w)-2/EEC(t)/T/SEC(b)-2/EWA(m)-2/Pz-6/Po-4/Pab-10/P1-4 IUP(c) AT

ACCESSION NR: AP5000833

\$ /0057/64/034/012/2114/2119

AUTHOR: Gott, Yu.V.; Tal'kovakiy, V.G.

TITLE: Analyzer for the energy spectrum of ions in high-temperature plasma

SOURCE: Zhurnal tekhnicheskoy fiziki, v.34, no.12, 1964, 2114-2119

TOPIC TAGS: plasma energy distribution, plasma diagnostics, ion analyzer

ABSTRACT: One of the significant characteristics of high-temperature plasmas, which is of particular interest in thermonuclear research, is the energy distribution of the ions. However, present techniques for determination of the ion spectrum optical procedures, high-frequency methods, ion traps and thin foils) all suffer from more or less serious shortcomings. Accordingly, there is proposed a technique and the design of an appropriate apparatus for ion spectrum measurements. The new departure is use of an ultrathin metal foil instead of a gas target, which makes possible substantial simplification of the installation. The analyzer is diagramed in Enclosure Ol. The neutral particles are incident on the ultrathin foil 1; the ions, forming incident to passage of the neutrals through the foil, are analyzed in energy in the field of the electrostatic capacitor 2. To increase the

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ACCESSION NR: AP5000833

angular aperture the capacitor plates have a special shape that allows of realizing double focusing of the ion beam. The foil window in the constructed analyzer measures 5 x 20 mm. The energy distribution is measured either by applying a saxtooth voltage to the capacitor plates to obtain a scan or at individual points, narrow energy groups being isolated by applying an appropriate dc bias from the rectifier 8 to the capacitor plates. After the capacitor, the ions pass through the aperture 11, are accelerated by the 5 kV potential difference and strike the first dynode of the electron multiplier 3; the multiplier signal is amplified by the wide-band amplifier 4 and either fed into the PS-10 000 scaler via the time discri instor 5 or applied to the plates of the oscillograph 6. A cross sectional view of the analyzer chamber is given in the figure. The choice of foil is discussed; silver foil 100-200 Å thick was used in the constructed analyzer. The said analyzer was used to investigate the distribution in energy of ions in an adiabatic trap with magnetic mirrors. The results are presented in the form of curves; the distributions obtained with fields of 5 and 8 kOe at the center of the trap are close and con-- story with the distribution determined otherwise. Particles with energies down to 40 Were recorded. It is noted that the reported measurements were intended primarily to verify the potentialities of the technique. Origiant has: 6 formulas and 7 figures.

L 47372=65 EPF(n)-2/EPA(w)-2/EWI(1)/EWG(m) P1-4/P0-4/P1-5/035-10 ACCESSION NR: AP5008736 s/0056/65/048/003/0804/0813 AUTHORS: Gott, Yu. V.; Yushmanov, Ye. Ye. TITLE: Experiments on heating of ions in magnetic mirror traps with the aid of an alternating electric field GOURCE: Zhurnal eksperimental'noy i teoreticheskoy fiziki. v. 48, 19. 3. 1965, 804-813 TOPIC TAGS: ion heating, plasma heating, magnetic trap, magnetic mirror, alternating field heating ABSTRACT: The authors present preliminary results of an investigation S plasma heating by a mod: fication of the so-calls: "ion magnetron" 191 described by the author (with M. S. lotre, ZhETF v. 1907. 1960 and v. 40, 40, 1961, and elsewhere). In this method I clasma beam is sent along the axis of a magnetic mirror dewise (probkotron), and a 20--30 usec pulsed potential difference of Cara 1/4

L 47372-65 ACCESSION NR: AP5008736 10--40 kV is applied between the plasma and the grounded walls. Alternating voltage injection was investigated, using the apparatus shown in Fig. 1 of the enclosure. The plasma source was a tungsten cathode, heated by an electron beam, and a tubular anode into which about 200 cc of hydrogen was fed continuously per hour. The plasma was studied by means of defectors of charge-exclusive particles (using constant electron emission), electric probes, and a spectrometer to fare the energies of the charge-exchange particles. All the instruments were located in the central cross-section plane of the trap. The plasma produced had a density ~1011 cm-3, the density of ions having energies larger than 5 keV being $\sim 3 \times 10^{10}$ cm⁻³. vacuum of $\sim 5 \times 10^{-7}$ mm Hg was maintained with the aid of flashed titanium. The electrons remained relatively cold. The plasma generated during a pulse decayed in about 150 μsec , mainly as a result of flute instability. Ion heating wis also observed at nonresonant frequencies. It is suggested that the ion acceleration mechanism has a stochastic character, although the main purpose of the work was to in-2/43

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ACCESSION NR: AP5008736			٤						
vestigate the possibilities of the described injection method, a preliminary interpretation of the results is in agreement with a theory that denies the possibility of pure cyclotron acceleration. The authors thank M. S. Loffe for suggesting the problem and for use- discussions, A. A. Smirnov for experimental assumed and Yallow and Yallow and I formula.									
ASSOCIATION: None									
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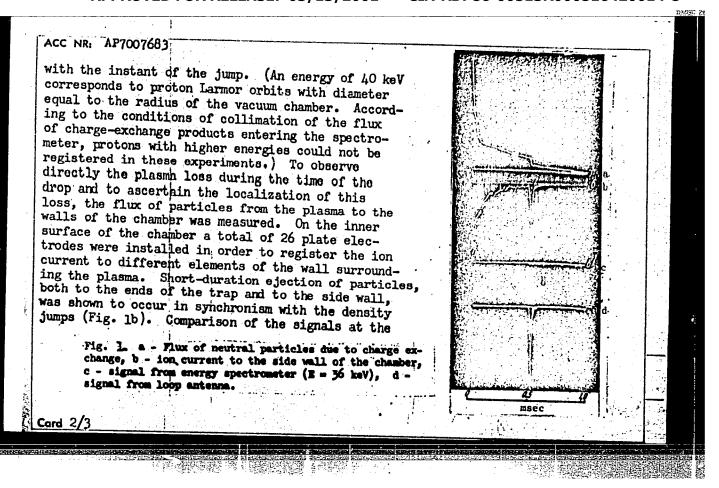
"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516420014-8

ACC NR AP7007683 SOURCE CODE: UR/0386/66/003/002/0092/0096 AUTHOR: Bayborodov, Yu. T.; Gott, Yu. V.; Ioffe, M. S.; Yushmanov, Ye. Ye. ORG: none TITIE: Unstable states of a plasma in a trap with combined field SOURCE: Zhurnal eksperimental'noy i teoreticheskoy fiziki. Pis'ma v redaktsiyu, v. no. 2, 1966, 92-96 TOPIC TAGS: plasma instability, plasma density, spectrometer, ion current, plasma ABSTRACT: The authors investigate distinct unstable states of a plasma observed in a trap with combined field ("minimum B" type) and offer a possible interpretation of the physical nature of the instability. It has been established that each density drop is accompanied by the appearance of high-frequency fields in the plasma. A loop antenna installed near the trap wall registered a burst of electromagnetic radiation whose spectrum consists of the ion-cyclotron frequency and its harmonics

(Fig. 1d); the frequency corresponds to the magnitude of the magnetic field in the central region of the trap. The burst duration, as well as the duration of the drop itself, is 15 - 20 usec. Figure lc shows the flux of neutral atoms produced by charge exchange and possessing an energy of 36 keV. At the start of the plasma

decay there are no ions with this energy, and their appearance coincides exactly **Card** 1/3



cyslotron instability is excited. The authors note that density jumps outward similar to those described in this article were observed also in a decaying plasma with hot electrons. In this case the instability develops at electron-cyclotron frequencies. Orig. art. has: 1 figure and 1 formula. SUB CODE: 20 / SUEM DATE: O2Dec65 / ORIG REF: 005 / OTH REF: 002	data shows that the ion-cyclotron the produced alter to high energies in apparently produced are produced in the	des showed that the loss occurs predominantly along the force ltant magnetic field through the end and radial mirrors. The he density jumps are due to a short burst of instability of type. This is evidenced both by the frequency spectrum of rnating fields and in the appearance of a group of ions acclerated in a transverse direction. The acceleration of the ions is a plasma and in this appearance of cyclotron frequency that	
Card 3/3	cyslotron instabil similar to those of with hot electrons frequencies. Orig	lity is excited. The authors note that density jumps outward lescribed in this article were observed also in a decaying plasma. In this case the instability develops at electron-cyclotron art. has: 1 figure and 1 formula.	
	Card 3/3		

COTTA, Alexandru, ing. tehnolog (Bucuresti)

Galculation of the micanite scale dimensions in manufacturing insultant cones for the collectors of the rotative electric machines. Electrotehnica 9 no.5:157-160 My *61.

1. Fabrica "Klement Gottwald", Bucuresti.

GOTTA, V.

"Logging and fish breeding in mountain streams." p. 41. (REVISTA PADURILAR, Vol. 68, no. 4, April 1953, Bucuresti, Rumania)

SO: Monthly List of East European Accessions, L. C., Vol. 3, No. 4, April 1954, Uncl.

GCTTE, A.

GCTTE, A. Survey of some fields of dressing of nonmetallic ores. p. 211.

No. 3, 1956.

RUPARSKO-METALURSKI ZBORNIK
TECHNOLOGY
Ljubljana, Yugoslavia

So: East European Accession, Vol. 6, No. 2, February 1957

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New data on the geology and ore potential of the Brandt region near Freiberg. Geol. rud. mestorosh. 5 no.2:111-114 Mr.Ap '63. (MIRA 16:6) 1. TSentral'nyy geologicheskiy institut Germanskoy Demokraticheskoy Respubliki. (Freiberg—Ore deposits)

RUMANIA

GOTTERBARM, P., MD, Pharmacist.

Bucharest, Farmacia, No 7, Jul 63, pp 435-442

"Stoichiometric Factors Useful in Pharmacoutical Practice."

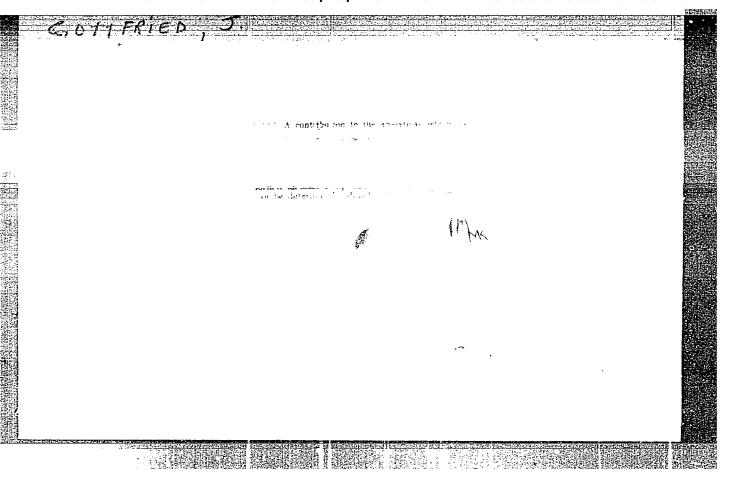
GOTTESHAN, Vil'yem L'vovich; YEFREMOVA, Ye.V., red.; ANDRIANOV, B.I., tekhn.

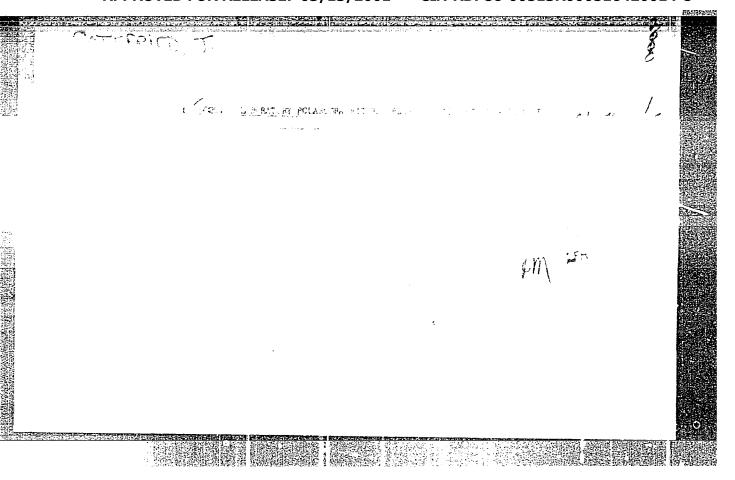
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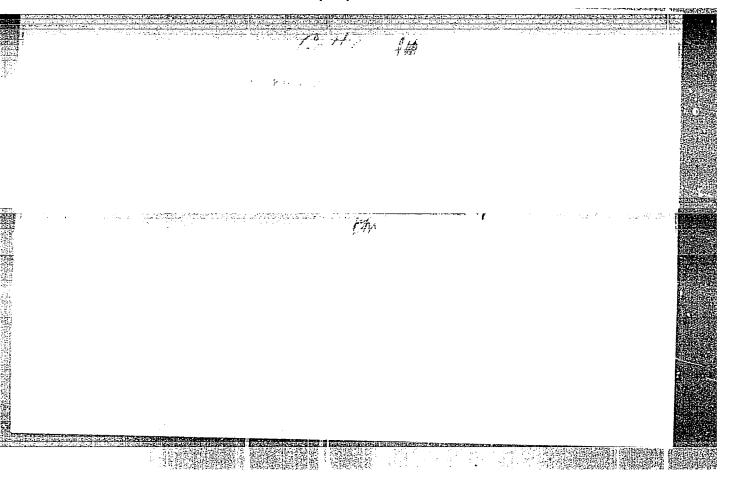
[Profiles for airplane models] Profili dlia letaiushchikh modelei.
Moskva, Izd-vo DOSAAF, 1958. 92 p. (MIRA 11:7)
(Airplanes--Models)

Transformation of a complete polygon to a star, p. 13. (PRZEGLAD ELEKTROTECHNICZNY, Warszawa, Vol. 31, no. 1, Jan. 1955.)

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, No. 15, Jan. 1955, Uncl.







CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57166.

Author : Shul'tsek Z., Gottfried J.

Inst : Not given.

Title : Rapid Determination Methods of Metals and of Raw

Materials. IV. Polarographic Determination of

Germanium.

Orig Pub: Chem listy, 1957, 51, No 11, 2010-2016.

Abstract: Germanium (Ge4+) when reduced from solutions at

pH >5, develops a peak in the emf curve. At peak value the voltage of -1.55 (compared to the saturated Hg2Cl2 electrode) can be attained on the

Card 1/4

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57166.

Abstract: background of Na₂CO₃, Na₂CO₃ and complexon III, or K₂CO₃ and KCN. The Ga wavelength increases with increased concentration of Na₂CO₃. The presence of small quantities of SiO₂ does not interfere with the determination of Ga. However, in the presence of large quantities of SiO₂ the Ga wave is completely suppressed. As the result of this, SiO₂Ch has to be removed beforehand by precipitation with caustic. NO₃ ions in contrast to SO₁Ch ions cause the distortion of the Ce waves. At high Cl concentrations the half wave potential of Ce wave is shifted toward more positive values. With increasing Cl concentration the peak of the Ce wave increases up to a limiting point and continues to

Card 2/4

14

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8 CZECHOSLOVAKIA / Analytical Chemistry: Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57166.

Abstract: remain at this value. The polarographic measurements, therefore, have to be conducted in a solution of approximately constant Cl concentration. The presence of V, and As(3+) interfere with polarographic determination of Ge. That of As(5+) does not. In the precipitation of Fe and Al hydroxides, Ge is adsorbed by the precipitates. Filtration of a solution through filter paper caused distortion and reduction of the polarographic wave of Ce. For the minerals of low Fe content, good results are obtained when alkaline fusion is employed, particularly when a fusion is performed with NaOH in the

Card 3/4

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2

Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 1, 1959, 989.

Author : Gottfried, J.
Inst : Not given.

: The Methods For Determining Titanium and Iron Title

in Clays Containing Titanium.

Orig Pub: Chem. prumysl., 1958, 8, No. 4, 176-180.

Abstract: The analytical results obtained by various meth-

ods on tatanium-containing clays are discussed; several thousand samples of various clays and intermediate products obtained in TiCl, production were used for the analyses. It was established that among the titrometric methods, the most suitable one is the method of potentiometric titration with CrSO4 [sic], which allows the sim-

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Card 3/3 APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8

CCUNTRY

: Czechoslovakia

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CATEGORY

: Analytical Chemistry.

ABS. JOUR.: ASKhim., No. 7, 1939, No. 23044

AUTHOR

INST.

: Sulcek, Z.; Sottfried, J.

TITLE

: Rapid Methods of Analysis of Metals and Mineral Haw Materials. IV. Polarographic Determination

of Germanium.

ORIG. PUB. : Collect. czechosl. chem. commun., 1958, 23,

No 8, 1515-1522

ABSTRACT

: See RZhKhim, 1958, 57166. Communication III

see RZhKhim, 1959, 947.

: CZECHOSLOVAKIA COUPTRY : Analytical Chemistry. Analysis of Enorganic CATEGORY Substances 1960, No. 845 : RZKhim., No. 1 ABS. JOUR. : Gottfried, J.; Jakovlev, J. V. AUTHOR : Determination of Copper in Metallic Germanium INST, TITLE by the Activation Rethod : Chem. prumysl, 1959, 9, 10 4, 179-182 ORIG. PUB. : The analyzed sample of metallic germanium was irrudiated in a nuclear reactor with a neutron flow of 5.1011-2.1013 neutrons/cm2 sec. The effectiveness of the chemical technique of separation and the degree of purification of ABSTRACT Cu from In was measured with the eid of labeled atoms. It was established that the admixture of In has practically no influence upon the accuracy of determination and that the separated 1/2CARD: E-14

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

GOTTFRIED, J.; JARA, V.

Determination of germanium in ammonium waters. p. 471.

CHEMICKY PRUMYSL. (ministerstvo chemickeho prumyslu) Praha, Czechoslovakia, vol. 9, no. 9, Sept. 1959.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, no. 11, Nov. 1959 Uncl.

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Z/009/60/000/04/007/041 E142/E235

5, 4500 (B AUTHORS:

Martinek, K., and Gottfried, J

Effect of Anions on B-Radiation of Potassium Salts TITLE:

PERIODICAL: Chemický průmysl, 1960, Nr 4, pp 192-194

ABSTRACT: Radioactive decomposition of the natural K40 isotope, proceeds at a very slow rate under formation of the nonradioactive isotopes Ar and Ca and is accompanied by β - and γ-radiation. It is, therefore, possible to assume that the isotope-composition of natural potassium is constant (Ref 1). In general, there is a linear dependence between the radiation intensity of potassium and its concentration in solutions (Ref 2), as well as in solid samples (Ref 3), except for potassium iodide (Ref 4) where the radiation intensity exceeds that due to the potassium content of the salt. It has been suggested that the increase in the radiation intensity of potassium salts in the presence of iodides and bromides is related to their concentration (Ref 6). Tests were carried out on the radiometric estimation of potassium in NPK fertilisers and the effect

of various anions on the β -radiation of potassium was Card 1/2 evaluated. It was found that the radiation intensity in

CIA-RDP86-00513R000516420014-8"

APPROVED FOR RELEASE: 03/13/2001

JAGER, Inbomir; GOTTFRIED, Jaroslav; NYVLT, Jaroslav.

Examination of the kinetics of crystalline urea drying. Chem prum 13 no.8:412-413 Ag 63.

1. Vyzkumny ustav anorganicke chemie, Usti nad Labem.

NYVLT, Jaroslav; GOTTFRIED, Jaroslav; KRICKOVA, Jaroslava

Control of the sintering tendency of crystalline urea. Pt. 12.

Chem prum 14, no.5:242-244. My '64.

1. Research Institute of Inorganic Chemistry, Usti nad Labem.

NYVLT, Jaroslav; GOTTFRIED, Jaroslav

Adaptation of the correlation method for calculation of solubility in ternary systems. Chem prum 14 no.7:376-378 Jl 164.

1. Research Institute of Inorganic Chemistry, Usti nad Labem.

NYVLT, J.; GOTTFRIED, J.; KRICKOVA, J.

On crystallization. Pt.10. Coll Cz Chem 29 no.1:161-167 Ja*64

1. Forschungsinstitut für anorganische Chemie, Usti nad Labem.

NYVLT, J.; GOTTFRIED, J.; KRICKOVA, J.

Crystallization. Pt. 11. Coll Cz Chem 29 no.10:2283-2289 0' 64.

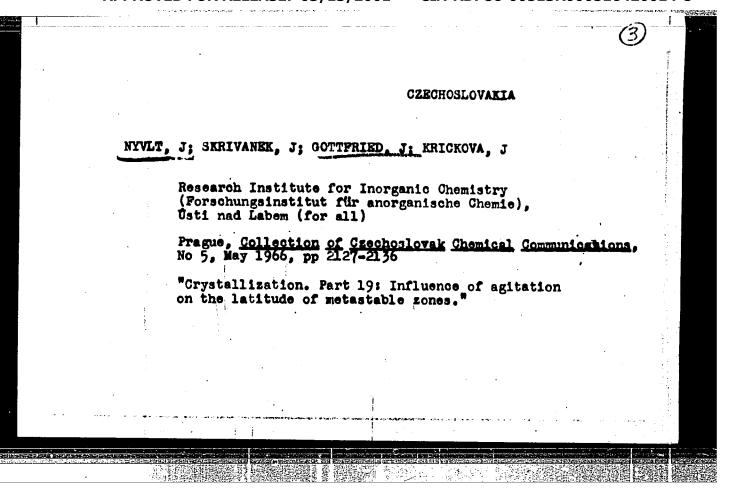
1. Forschungsinstitut fur anorganische Chemie, Usti nad Labem.

JAGER, Lubomir; GOTTFRIED, Jaroslav; NYVLT, Jaroslav; SURA, Jindrich

Kinetics of forming biuret from urea. Chem prum 15 no.1: 4-7 Ja 165.

1. Research Institute of Inorganic Chemistry, Usti and Labem.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

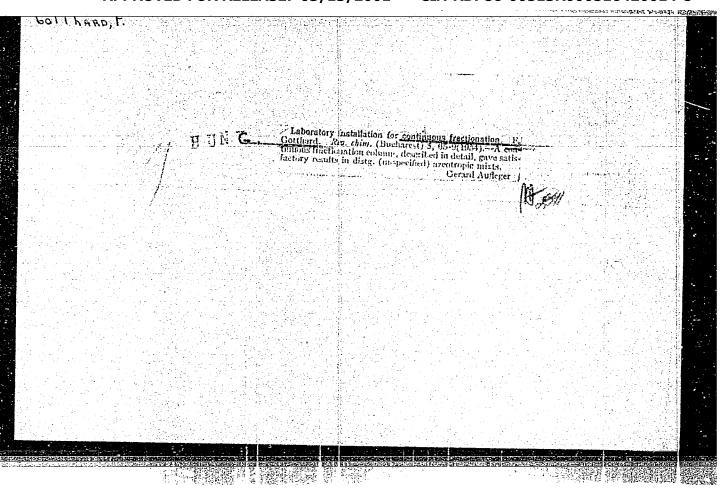


ACCESSION NR: AP5020840	cz/0034/64/000/009/0609/0617 / 7
AUTHOR: Harok, Kilan (Engi	neer); Gottfried, Kamil (Engineer); Hanza, Zdenek B
3	dized in the ingot mold. Part I.
SOURCE: Hutnicke listy, no	9, 1964, 609-617
TOPIC TAGS: steel, metallu	urgic furnace, metal casting
pleted, and remaining the mold deoxidized standard layer, and the of the teeming rate. I as the killed steel grives better surface of	Itsh summary 7: Balanced steel deoxidized anufactured In the gent was a furnace by remaining steel. Amount of Al used as deoximunt of C in the heat. The best was of j snortly effor the length thering In tom-2/3 shortly after that. The structure of sel ingot consists of the balanced steel balanced core. The structure is a function he steel is nearly as chemically homogenous ades. Bottom-poured, mold-deoxidized steel f the rolled products, and higher yields brig. art. has 3 formulas, 9 graphs, and 6 tables.
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ASSOCIATION: Institute, NE	Vyskumu y a sk	usební ustav	NHKG, Kuncice (Research and	Control	
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	E/P(b)/E/T(t) R: AP5020871	CZ/0034/64/000/010/0693/0698	/3 /3
AUTHOR: He	rok, Hilan (Engineer);	Gottfried, Kamil (Engineer); Hamza, Z. (Enginee	<u> </u>
TITLE: Bal	anced steel deoxidized	I in the inget gold. Part II. Properties of stee	1
SOURCE: Ru	tnicke listy, no. 10;	1964, 693-698	
TOPIC TAGS:		metal test, tensile stress, pipe, fabricated	
billete. Di med* of bal showed that substitute.	scussed is the scatter ance steel, and minimu ourd. A comparison of balanced steel resemb	determine the segregation degree of balanced stering of values in tensile tests of tubes and strium segrebation and dispersion of the tensile test transition temperatures, aging, and weldability bled rimming steel grades, and may be used as its ed steel only in exceptional cases. Orig. art.	.ps ,
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RUMANIA/Chemical Technology. Chemical Products and Their

Application. Part 3. - Industrial Organic

Synthesis.

Abs Jour: Referat. Zhurnal Khimiya, No 21, 1958, 71643.

Author : F. Gotthard, Martha Gotthard.

Inst Title

: Effect of Distillate Yield on Separation of Components of N. Hexane, Methylcyclopentane and Benzene Mixture at Extractive Distillation with

Ariline.

Orig Pub: Rev. chim., 1956, 7, No 11, 634-638.

Abstract: Results or experimental extractive separation

with aniline of a mixture containing (in 4% by weight) cyclopentane - 1.5, 2-methylpentane - 33.3,

n-hexane - 33.2, methylcyclopentane - 19, and ben-

: 1/3 Card

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8

RUMANIA/Chemical Technology. Chemical Products and Their Application, Fart 3. - Industrial Organic Synthesis.

Abs Jour: Referat. Zhurnal Khimiya, No 21, 1958, 71643.

zene - 13 using a laboratory distillation column of 33 theoretical plate productivity are presented. The yield of naphthenes (I) drops sharply at the distillate amount above 65%. The concentration of I in the residue is maximum (53 to 55%) at the distillate yield of 73 to 75%, after which it sharply drops; the residue still contains considerable amounts of paraffins (II) (from 3 to 29.6%) at the distillate yield of 80%. The best recommended technological scheme of methylcyclopentane separation foresses a two-stage extractive distillation: the first column works with a distillate yield below 73 to 75% in order to secure a large yield of I in

: 2/3 Card

RUMANIA/Chemical Technology. Chemical Products and Their Application, Part 3. - Industrial Organic Synthesis.

Abs Jour: Referat. Zhurnal Khimiya, No 21, 1958, 71643.

the residue, which undergoes a second extractive distillation. A heavy fraction without I and a light fraction containing benzene and I besides II are obtained with the second column. The light fraction is recirculated into the first column.

Card: 3/3

72

RUMANIA / Chomical Technology. Chemical Products H-23 and Their Applications. Chemical Process.

APPROVED FOR RELEASE 193/13/2991 and FJATE 193/13/20014-8' and Rocket Fuel Lubricants.

Abs Jour: Ref Zhur-Khimiya, No 3, 1959, 9710.

Author : Gotthard, F. Inst : Not given.

Title : Separation of Cyclohexane from Benzene by Rec-

tification.

Orig Pub: Rov. chim., 1957, 8, No 10, 644-649.

Abstract: The possibility of separating cyclohoxane (I) from benzene (II) by rectification depends on the chemical composition of II and is determined chiefly by the ratio of C₆H₆ (III) concentration and the other components in the basic fractions of raw material. I is separated by the primary

Card 1/4

RUMANIA / Chemical Tochnology. Chemical Froducts H-23 and Their Applications. Chemical Processing of Natural Gases and Petroleum. Notor and Rocket Fuel Lubricants.

Abs Jour: Ref Zhur-Khimiya, No 3, 1959, 9710.

Abstract: and IV; the liborated I is contaminated only with a small quantity of V. c) II contains a largor quantity of III, sufficient either for partial or for total removal of I in the form of A; I in the fraction 79-820 is badly contaminated by V and 2,2,3-dimothylbutane or is not contained in this fraction at all. A three dimensional diagram is given, to determine the possibility of producing I and its yield in rectification of the fraction with a boiling point of 72-820 depending on the content of I, III and IV in this fraction. On a laboratory column (53 theoretical plates) with an average fractional

Card 3/4

189

GOTTHARD, F.; REUSZ, N.

Testing a pellicular evaporator. p. 204.

REVISTA DE CHIMIE. Bucuresti, Rumania. Vol. 10, no. 4, Apr. 1959.

Monthly List of East European Accessions. (EEAI), LC. Vol. 8, no. 9,/1959. Uncl.

RNAME, Given Names	
Country: Rumania	
Academic Degrees:	
Affiliation: -not given-	
Source: Bucharest, Revista de Chimi	e, Vol 12, No 8, Aug 1961, pp 489-491.
The Arman Control of the Control of	e-Bensene Mixtures and the Idquid-Vapor
Equilibrium or these Mixtures	•"
Authors:	
GOTTHARD, Fr., -Engineer	
MINEA, I., -Engineer	
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GOTTHARD, Fr.; MINEA, I.

Separation of methyl-cyclopentane from benzene by rectification.

II. Rev chimie Min petr 13 no.3:134-140 Mr '62.

ACCESSION NR: AP4039548

R/0003/64/015/005/0252/0256

AUTHOR: Ciocoiu, Paulina; Gotthard, Fr.; Minea, I.; Russu, R.

TITLE: Synthesis and applications of some molecular sieves. I. Some properties and uses of molecular sieves

SOURCE: Revista de chimie, v. 15, no. 5, 1964, 252-256

TOPIC TAGS: molecular sieve, petrochemical application, persorption, aluminumsilicates, crystalline structure, geometrical selectivity, physical selectivity, persorption heat, purification, separation, pH, hydrocarbons, mercaptan, drying, monomers, inert gas, Ar, H sub 2S, H sub 2, O sub 2, N sub 2, C sub 2, H sub 4, ,

ABSTRACT: Molecular sieves are rigid substances with tridimentional structure having pores and cavities which allow selective penetration of atoms or molecules of different dimensions. This first note is a short review of literature on the subject with emphasis on application on molecular sieves to petrochemical industry.

ASSOCIATION: Institutul Petrochim Ploiesti (Ploiesti Petrochomical Institute)

Card 1/2

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GOTTHARD, J.

The Kaldska Peat Bog, state wildlife reservation. p. 116. CCHRANA PRIRODY. (Ministerstvo kultury. Statni pece o ochranu prirody) Praha.
Vol. 11, no. 4, May 1956.

SCURCE: EEAL - LC Vol. 5 No. 10 Oct. 1956

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

HUNGARY

KRAKOVITS, Gabor, SOREL, Matyas, and GOTTHARD, Lajos: Department of Orthopedics (Orthopaediai Osztaly) of the Janos Hospital (Janos Korhaz) [Budapest].

"Sexual Differences in the Hip-Joint Flexibility of New-Born Animals"

Budapest, Kiserletes Orvostudomany, Vol 18, No 6, 1966; pp 600-603.

Abstract: Using the method described by Crelin, the authors demonstrated the existence of sexual differences in the hip-joint flexibility of new-born animals. In the case of female animals (rats, hamsters, guinea pigs) the flexibility of the hip joint is greater, even from a physiological viewpoint, than in male animals. These findings are related to the fact that the incidence of hip sprain or dysplasia in young girls is higher than in boys. The sexual difference in the flexibility of the hip joint is attributed, on the basis of literature data, to a hormonal effect. 15 References, all Western. Manuscript received

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GOTTHARD, MARTHA

RUMANIA/Chemical Tachnology. Chemical Products and Their

Application. Part 3. - Industrial Organic

Synthesis.

Abs Jour: Referat. Zhurnal Khimiya, No 21, 1958, 71643.

Author : F. Gotthard, Martha Gotthard.

Inst

Title : Effect of Distillate Yield on Separation of Components of N-Hexane, Methylcyclopentane and Benzene Mixture at Extractive Distillation with

Ariline.

Orig Pub: Rev. chim., 1956, 7, No 11, 634-638.

Abstract: Results of experimental extractive separation

with aniline of a mixture containing (in 5% by weight) cyclopentane - 1.5, 2-methylpentane - 33.3, n-hexane - 33.2, methylcyclopentane - 19, and ben-

Card : 1/3

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RUMANIA/Chemical Technology. Chemical Products and Their Application, Fart 3. - Industrial Organic Synthesis.

Abs Jour: Referat. Zhurnal Khimiya, No 21, 1958, 71643.

zene - 13 using a laboratory distillation column of 33 theoretical plate productivity are presented. The yield of nephthenes (I) drops sharply at the distillate amount above 65%. The concentration of I in the residue is maximum (53 to 55%) at the distillate yield of 73 to 75%, after which it sharply drops; the residue still contains considerable amounts of paraffins (II) (from 3 to 29.6%) at the distillate yield of 80%. The best recommended technological scheme of methylcyclopentane separation foresses a two-stage extractive distillation: the first column works with a distillate yield below 73 to 75% in order to secure a large yield of I in

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Hottich, B. P.

USSR/Organic Chemistry. Synthetic Organic Chemistry. **E-2**

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26699.

Kochetkov, N.K., Khorlin, A.Ya., Gottikh, B.P.; Nesmeyanov, A.N. Academy of Sciences of USSR. Author

Synthesis of Alkenyl- \$\beta\$-chlorovinylketones. Inst Title

Izv. AN SSSR, Otd. khim. n., 1956, No. 9, Orig Pub

1053 - 1058.

Abstract :

The following methods of synthesis of alkenyl-B-chlorovinylketones (ACVK) were developed: the method of direct condensation of acetylene (I) with chloroanhydrides (CA) of α, β-unsaturated acids in presence of AlCl3, and the method
of condensation of I with CA of α- or β-chlororeplaced acids in presence of AlCl3 with follow-

ing dehydrochlorination of the produced a-

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USSR/Organic Chemistry. Synthetic Organic Chemistry. E-2

Abs Jour: Ref Zhur - Khimiya, No. 8, 1957, 26699.

(or B)-chloroalkyl-B-chlorovinylketone. Experiments with CA of C- or B-bromoreplaced periments with CA of C- or B-bromoreplaced acids did not succeed. It is shown that ACVK acids did not succeed. It is shown that ACVK react with B-naphthol (II) in presence of the color of the soliton point 115 to 1500 is fraction of the boiling point 115 to 1500 is collected, shaken with mercury, distilled using a column still (10 to 15 theoretical using a column still (10 to 15 theoretical (III) is 60.3%, boiling point 129 to 1320/

756 mm, n²⁰D = 1.4475, d₄ 17 = 1.2360. CA of

Card 2/6

USSR/Organic Chemistry. Synthetic Organic Chemistry. E-2

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26699.

isovaleric acid, 167 g of SO₂Cl₂ and 1 g of I₂, the yield was 58%, boiling point 145.5 to 146.50/ the yield was 58%, boiling point 145.5 to 146.50/
749 mm. I is conducted into a solution of 81 g
of CA of \$\mathcal{B}\$-chloropropionic acid (V) in 150
of CA of \$\mathcal{B}\$-chloropropionic acid (V) in 150
mlit of CClh (6 hours) and 85 g of AlCl3 is
mlit of CClh (6 hours), at 10 to 150), poured
introduced (2 hours), at 10 to 150), poured
out on ice, CHCl3 is extracted with 63 g of
out on ice, CHCl3 is extracted with 63 g of
out on ice, CHCl3 is extracted with 63 g of
a mixture of vinyl \$\mathcal{B}\$-chlorovinylketone (VII),
and \$\mathcal{B}\$-chloroethyl \$\mathcal{B}\$-chlorovinylketone (VII),
and \$\mathcal{B}\$-chloroethyl \$\mathcal{B}\$-chlorovinylketone (VIII),
at treated with 65 g of C6H5N(C2H5)2 (VIII) at
treated with 65 g of C6H5N(C2H5)2 (VIII) at
treated with 65 g of C6H5N(C2H5)2 (VIII) at
treated with 65 g of C6H5N(C2H5)2 (VIII) at
treated with 5% H2SO4; the
ether, filtered, washed with 5% H2SO4; the
yield of VI is 31.5%, boiling point 48 to 49.50/

 $14 \text{ mm}, n^{20}D = 1.4938, d_{4}^{20} = 1.1274.$ 57 g of a

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APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

USSR/Organic Chemistry. Synthetic Organic Chemistry. E-2
Abs Jour: Ref Zhur - Khimiya, No. 8, 1957, 26699.

mixture of propenyl. 8 -chlorovinylketone (IV) and 6-chloropropyl. 8-chlorovinylketone, boiling point 81 to 860/15 mm, was received analogously of 81 g of III and 85 g of AlCl3 in 200 mlit of CCl4 by conducting I (7 to 8 hours at 15 to 200); after the treatment with boiling point 70 to 720/10 mm, melting point 38 to 390. A fraction of the boiling point of 78 to 920/10 mm was received in the amount of 30 g from 60 g of IV and 60 g of AlCl3 ing I (7 hours at 15 to 200); it was treated with 32.5 g of VIII, yield of isobutenyl. 8- chlorovinylketone (XI) was 32.7%, boiling point 76 to 790/11 mm. 13.3 g of AlCl3 is added to

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USSR/Organic Chemistry. Synthetic Organic Chemistry. E-2
Abs Jour: Ref Zhur - Khimiya, No. 8, 1957, 26699.

10.5 g of CA of crotonic acid in 50 mlit of X, and I is let through (3.5 hours at 20 to 250), X is separated, extracted with CHCl3, yield of IX is 50.5%, boiling point 64 to 650/7 mm. XI (yield 34%, boiling point 71 to 730/

10 mm, $n^{20}D = 1.5038$, $d_{l_{+}}^{20} = 1.0644$) was obtained

analogously from 25 g of CA of B, B-dimethylacrylic acid in 75 mlit of X and 32 g of AlCl3 by letting through I (9 hours at 8 to 120).

60 g of AlCl3 is added (2 hours) to a solution of 57.3 g of V in 120 mlit of CCh., I is let through (7 hours at not above 20°), the yield of VI is 17.7%; the yield of VII is 38.5%

boiling point 100 to $102^{\circ}/15$ mm, $n^{20}D = 1.50+5$,

Card 5/6

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

USSR/Organic Chemistry. Synthetic Organic Chemistry. E-2
Abs Jour: Ref Zhur - Khimiya, No. 8, 1957, 26699.

 $d_{4}^{20} = 1.2754$. 2 g of II and 12 mlit of water-

free FeCl₃ in concentrated HCl is added to a solution of 1.3 g of IX in 20 mlit of glacial CH₃COOH. The yield of ferrichloride of 2-propenylnaphthowl, 2:5,6-pyrilium is 67%, melting point 135 to 137° (from glacial CH₃COOH, dissoc.). Ferrichloride of 2-losbutenylnaphthowl, 2:5,6-pyrilium was prepared analogously of 0.7 g of XI in 10 mlit of glacial CH₃COOH and 1 g of II and 6 mlit of FeCl₃ in concentrated HCl; yield 71.5% melting point 186 to 188° (from glacial CH₃COOH).

Card 6/6

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

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ARENDARUK, A.P.; BUDOVSKIY, B.I.; GOTTIKH B.P.; KARPEYSKIY, M.Ya.

KUDRYASHOV, L.I.; SKOLDINOV, A.P.; SMIRNOVA, N.V.; KHORLIN, A.Ya.

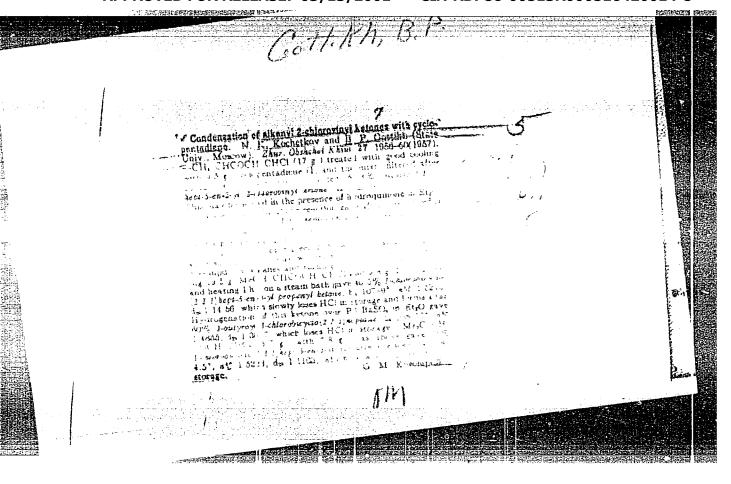
KOCHETKOV, N.K.

Dihydrosarcomycin and related compounds. Part.1. Zhur.ob.khim.
27 no.5:1312-1318 My '57.

(NLRA 10:8)

1.Institut farmakologii i khimioterapii Akademii meditsinekikh
nauk SSSR.

(Antibiotics)



"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516420014-8

sov/63-3-6-36/43

· Kochetkov, N.K., Gottikh, B.P., Karpeyskiy, M.Ya., Khomutov, R.M. AUTHORS:

The Configuration of eta -Chlorovinylketones (O konfiguratsii TITLE:

β-khlorvinilketonov)

Khimicheskaya nauka i promyshlennost', 1958, Vol III, Nr 6, PERIODICAL:

p 834 (USSR)

It is supposed that $oldsymbol{eta}$ -chlorovinylketones have a trans-con-ABSTRACT:

figuration, since the only product of the oxidation of the sodium hypochlorite of the methyl- β -chlorovinylketone is the

trans-chloroacrylic acid.

There are 6 Soviet references.

ASSOCIATION: Nauchno-issledovatel'skiy institut farmakologii i khimoterapii

(Scientific Research Institute of Pharmacology and Chemical

Therapy)

May 7, 1958 SUBMITTED:

Card 1/1

CIA-RDP86-00513R000516420014-8" APPROVED FOR RELEASE: 03/13/2001

AUTHORS:

SOV / 79-28-6-15/63 Kochetkov, N. K., Gottikh, B. P., Kudryashov, L. I.

TITLE:

The Conversion of $\mathcal B$ --Chlorovinylketones With $\mathcal B$ -Dicarbonyl Compounds (Vzaimodeystviye β -khlorvinilketonov s β -dikarbonil nymi soyedineniyami) V. Ketovinylation of a-Alkyl Acetoacetic Esters (V. Ketovinilirovaniye -alkilatsetouksus-

nykh efirov)

PERIODICAL:

Zhurnal obehchey khimii, 1958, Vol. 28, Nr 6, pp. 1508-1511

(USSR)

ABSTRACT:

In continuation of the previous papers on the synthesis of Y-ketcalkenylmalonic esters it was of interest to the authors to extend the limits of ketonylation of the compounds with a movable methylene-and methenyl member. First the authors investigated the ketovinylation of the a-alkylacetoacetic esters described in this paper. Based on the experience collected (Ref 4) on the conversion of the β -chlorovinylketones with acetoacetic ester in the presence of potash in boiling toluene, in the presence of only one movable hydrogen atom, secondary processes could be excluded in the present case and

Card 1/3

SOV/79-28-6-15/63

The Conversion of β -Chlorovinylketones With β -Dicarbonyl Compounds. W. Ketovinylation of A-Alkyl Acetoacetic Esters

the sole formation of products of a normal ketovinylation could be expected. In the conversion of β —chlorovinylketones with sodium derivatives of the λ —alkyl acetoacetic esters in benzene the hitherto unknown λ —alkyl— λ —(γ —ketoalkenyl)—acetoacetic esters were actually obtained without difficulty and in good yields:

RCOCH=CHCl+CH₃COCHR'COCC₂H₅ \rightarrow RCOCH=CHC-COCC₂H₅ $R^{\bullet}=CH_3, C_2H_5, C_3H_7$ $R^{\bullet}=C_2H_5, C_4H_9$ COCH

Owing to the accessibility of the initial substances and its simple method this reaction is a convenient method for the synthesis of these interesting and much promising compounds. They are high-boiling and stable oils which, however, easily resinify in distillation in an insufficient vacuum. Their structure is proved analytically. The result of the hydration of one of these compounds points to the presence of a double bond in the ketovinylation products, which fact is to be proved further by conversions. There are 5 references, 5 of which are Soviet.

Card 2/3

SOV/79-28-6-15/63

The Conversion of β -Chlorovinylketones With β -Dicarbonyl Compounds. W. Ketovinylation of 2-Alkyl Acetoacetic Esters

ASSOCIATION: Institut farmakologii i khimioterapii Akademii meditsinskikh

nauk SSSR

(Pharmacological and Chemo-Therapeutic Institute, Academy of

Medical Sciences USSR)

SUBMITTED: May 27, 1957

1. Ketones-Chemical reactions

Card 3/3

AUTHORS:

Kochetkov, N. K., Gottikh, B. P.

SOV/79-28-10-23/60

TITLE:

Reaction of β -Chloro-Vinyl Ketone With β -Dicarbonyl Compounds (Vzaimodeystviye β -khlorvinilketonov s β -di-karbonil'nymi soyedineniyami) VII. Acid Cleavage of the α -Alkyl- α -(3-Ketoalkenyl)-Acetic Esters (VII. Kislotnoye rasshchepleniye α -alkil- α -(1-ketoalkenil)-atsetouksusnykh

efirov)

PERIODICAL:

Zhurnal obshchey khimii, 1958, Vol 28, Nr 10,

pp 2732 - 2735 (USSR)

ABSTRACT:

The α -alkyl- α -(3-ketoalkenyl)-acetic esters obtained by the authors (Ref 1) by the keto vinylation of α -alkyl acetic ester are of interest for organic synthesis. The authors began their investigation with the reaction of their acid cleavage. These compounds are

related to the acid decomposition of α , α -diacyl derivatives of the esters of fatty acids so that an easier separation of the acyl group than of the β -ketovinyl group was to be expected. The cleavage of the α -alkyl- α -(3-ketoalkenyl)-

acetic esters according to Dieckmann (Ref 2) and Bouveault (Ref 3) (Dikman, Buvo) failed due to their

Card 1/3

Reaction of β -Chloro-Vinyl Ketone With β -Dicarbonyl SOV/79-28-10-23/60 Compounds. VII. Acid Cleavage of the α -Alkyl- α -(3-Ketoalkenyl)-Acetic Esters

tendency to complex condensations and to the formation of side products. The use of aqueous ammonia solution with the addition of ammonium chloride (Ref 4) was successful in this cleavage, with the above mentioned esters being subjected to a sufficiently easy acid cleavage under the formation α -(3-ketoalkenyl)-alkane acids (See Reaction Scheme 1). The more accurate testing of this reaction by the decomposition of α -ethyl- α -(3-ketobutenyl)-acetic ester showed that the final result depends on the amount of the related ammonia solution, as it may be seen from the data given in table 1. The reaction is of general character. It is shown that the acid cleavage with the increase of the molecular weight of the α -alkyl- α -(3-ketoalkenyl)-acetic esters demands stricter conditions and at the same time offers smaller yields of α -(3-ketoalkenyl)-alkane acids. There are 1 table and 6 references, 2 of thich are Soviet.

Card 2/3

Reaction of β -Chlore-Vinyl Ketone With β -Dicarbonyl 80V/79-28-10-23/60 Compounds. VII. Acid Cleavage of the a-Alkyl-a-(3-Ketoalkenyl)-Acetis Esters

ASSOCIATION: Institut farmakelegii i khimioterapii Akademii meditsinskikh

nauk SSSR (Institute of Pharmacology and Chemotherapy of the Academy of Medical Sciences USSR)

September 30, 1957 SUBMITTED:

Card 3/3

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516420014-8"

COTTIKH, B. P.: Master Chem Sci (diss) -- "Ketovinylation of the esters of betaketo acids and beta-diketones". Moscow, 1959. 8 pp (Moscow Order of Lenin and Order of Labor Red Banner State U im M. V. Lomonosov), 100 copies (KL, No 10, 1959, 123)

5(3) SOV/79-29-4-60/77

AUTHORS: Kochetkov, N. K., Gottikh, B. P., Shtumpf, Rol'f

TITLE: Reaction of β -Chlorovinyl Ketones With β -Dicarbonyl Compounds

(Vzaimodeystviye β -khlorvinilketonov s β -dikarbonil'nymi soyedineniyami). IX. Ketovinylation of the Esters of Cyclic β -Keto

Acids (IX. Ketovinilirovaniye efirov tsiklicheskikh β-ketokislot)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 4, pp 1320-1323 (USSR)

ABSTRACT: In connection with previous investigations carried out by the

authors (Refs 1-5) the ketovinylation of the esters of cyclic keto acids is discussed in the article under review, and new prospects for their synthesis are given. Subject of the examination were the ethyl esters of cyclopentanone- and cyclohexanone carboxylic acids containing active hydrogen and therefore capable of being easily ketovinylated (Ref 6). The ketovinylation of the

said β -keto esters was carried out by reaction of the sodium

derivative of the β -keto ester suspended in benzene with β -chlorovinyl ketone, according to references 2 and 3. The reac-

tion, which met with no difficulties whatever, neither in the case of alkyl β -chlorovinyl ketones, nor in the case of their

Card 1/3 aromatic analogues, resulted in normal ketovinylation products

\$50V/79-29-4-60/77\$ Reaction of $\beta-Chlorovinyl$ Ketones With $\beta-Dicarbonyl$ Compounds. IX. Ketovinylation of the Esters of Cyclic $\beta-Keto$ Acids

with yields ranging from 60 to 70% (Scheme 1). After the usual treatment the 1-carbethoxy-1-(3'-ketoalken-1'-yl-1')-cycloalkanones-2 were removed by distillation (see the experimental part). In principle, this reaction does not differ from the ketovinylation of α -alkylacetacetic esters (Ref 3). In order to realize the ketovinylation of other β-dicarbonyl compounds of the alicyclic series, one of the representatives of cyclic β -diketones, viz. 1,1,4-trimethylcyclohexanedione-3,5 (methyldimedone) was ketovinylated. After many, partly unsuccessful, experiments the conditions were found under which it is possible to prepare a ketovinylation product with an output of between 30 and 35% (Scheme 2). When hydrogenated this product absorbs 1 mol of hydrogen, which suggests only one double bond, and does not yield a 1,3,5-triacetyl benzene with diluted acids, which suggests a C derivative. These results confirm the theory of the structure of the product obtained and prove the fact that it is possible to apply ketovinylation to cyclic as well as alicyclic β-diketones. There are 6 Soviet references.

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SOV/79-29-4-60/77

Reaction of β -Chlorovinyl Ketones With β -Dicarbonyl Compounds. IX. Ketovinylation of the Esters of Cyclic β -Keto Acids

ASSOCIATION:

Institut farmakologii i khimiotorapii Akademii meditsinskikh nauk SSSR (Institute of Pharmacology and Chemotherapy of the

Academy of Medical Sciences, USSR)

SUBMITTED:

March 6, 1958

Card 3/3

5(3) AUTHORS:

Kochetkov, N. K., Gottikh, B. P. Vinokurov, V. G., khomutov, R. M.

SOV/20-125-1-23/67

TITLE:

On the Structure of A.Chlorovinyl Ketones and on the Stereochemistry of the Reaction of Ketovinylation (O konfiguratsii A-khlorvinilketonov i stereokhimii reaktsii

ketovinilirovaniya)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 1, pp 89-92

ABSTRACT:

The structure of the substances mentioned in the title RCOCH=CHCl is, in spite of their well elaborated utilization methods (Ref 1), still an unsolved problem. From the most important methods of production (Refs 2-4) it may be assumed that the substances produced in this way have a trans-structure. The authors succeeded in clearly confirming experimentally this assumption. If one of the simple β -chlorovinyl ketones,

methyl-\beta-chlorovinyl ketone is oxidized with sodium

hypochlorite, the trans-f-chloro acrylic acid (Ref 5) forms under rigidly controllable conditions as the only product. If this oxidation does not contact the C-atoms with a multiple binding, moreover, if the mild conditions of reaction exclude

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On the Structure of β -Chlorovinyl Ketones and SOV/20-125-1-23/67 on the Stereochemistry of the Reaction of Ketovinylation

the iscmerization of the initial substance and the reaction product a complete transformation of the structure during the reaction is impossible. Due to this fact methyl- β -chlorovinyl ketone has to be regarded as a transisomer. Thus, also all alkyl-, alkenyl-, and aryl-p-chlorovinyl ketones (Refs 2-4) are transisomers under similar conditions. As far as the β -chlorovinyl ketones (Refs 6, 7) produced by other methods are identical with those obtained by condensation with acetylene, they are obviously also transisomers. By the knowledge of the above structure the stereochemistry of the reaction mentioned in the title (Ref 1) could be observed. It is one of the most important reactions of eta -chlorovinyl ketones and is only a mucleophilic substitution of a halogen atom. Since the chemical methods cannot be used for determining the structure of the reaction products mentioned the authors used infra-red spectra. Although the authors mention only data on the ketovinylation of sulfinic acids and eta -dicarbonyl compounds, they have little doubt that also in other cases (Ref 1) ketovinylation reaction leads to a formation of transisomers. In other words, the reaction takes place under

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On the Structure of \(\beta \) -Chlorovinyl Ketones and SOV/20-125-1-23/67 on the Stereochemistry of the Reaction of Ketovinylation

preservation of the structure of the keto-vinyl group of the initial \$\beta\$-chlorovinyl ketone. This preservation may be explained by the substitution mechanism of the halogen (Ref 1, see Scheme) suggested by the author mentioned first. There are 3 figures and 16 Soviet references.

ASSOCIATION:

Institut farmakologii i khimioterapii Akademii meditsinskikh nauk SSSR (Institute of Pharmacology and Chemotherapy of the Academy of Medical Sciences, USSR)

PRESENTED:

December 1, 1958, by A. N. Nesmeyanov, Academician

SUBMITTED:

November 29, 1958

Card 3/3

5.3400

78288 SOV/79-30-3-42/69

AUTHORS:

Kochetkov, N. K., Gottikh, B. P.

TITLE:

Reaction of β -chlorovinyl ketones With β -Carbonyl Compounds. XI. Ketovinylation of Methylacetylacetone and 2-Methyldlhydroresorcinol. Synthesis of Un-

saturated 8-Diketones

PERIODICAL:

Zhurnal obshchey khimi, 1960, Vol 30, Nr 3, pp 948-953 (USSR)

ABSTRACT:

The reaction of the sodium derivative of methylacetone with G -chlcrovinyl ketones in benzene, yields the following ketovinylation products: methy1-(3-ketobuten-1-y1-acetylacetone (I), yield 59%, bp $100-101^{\circ}$ (1 mm), n_{D}^{20} 1.4860; methy1-(3-

ketopent-1-y1)-acetylacetone (II), yield 44%, bp

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106-107.5° (1 mm), n_D^{20} 1.4822; methyl-(3-ketohexen-1-yl)-acetylacetone (III), yield 59%, bp 116-118° (1 mm),

heaction of β -Chlorovinyl Ketones With β -Carbonyl Compounds. XI 78288 sov/79-30-3-42/69 1.4795. Ketovinylation of 2-methyl-dihydroresorcinol in dioxane yields 2-methyl-2-(3-ketobut-1-yl)--dihydroresorcinol (IV), yield 26%, mp 61-63.5,0 previously not known. CH=CHCOR (I) $R = CH_{1i}$ (II) $R = C_1H_{1i}$ (III) $R = C_2H_{1i}$ + CH₃COCH=CHCl -NaCl 'СИ≕СИСОСН_а Methyl-(3-ketoalk-1-yl)-acetylacetones yield un-S-diketones when subjected to alkali Card 2/5 treatment.

Reaction of β -Chlorovinyl Ketones With β -Carbonyl Compounds. XI

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 $\begin{array}{c} \mathrm{CH_3COC}(\mathrm{CH_3})\mathrm{COCH_3} & \xrightarrow{\mathrm{OIP}} & \mathrm{RCOCH} = \mathrm{CHCH}(\mathrm{CH_3})\mathrm{COCH_3} \\ \\ \mathrm{CH} = \mathrm{CHCOR} \\ \\ \mathrm{(V)} & \mathrm{R} = \mathrm{CH_3}, & \mathrm{(VI)} & \mathrm{R} = \mathrm{C_1H_4}, & \mathrm{(VII)} & \mathrm{R} = \mathrm{C_1H_4}. \end{array}$

The following unsaturated δ -diketones were prepared: 3-methylhept-4-ene-2.6-dione (V), yield 76.5%, bp 74-75.5° (1 mm), n20 · 1.4756; 3-methyloct-4-ene-2.6-dione (VI), yield 76.5% bp 82-83° (1 mm), nD 1.4743; 3-methylnon-4-ene-2.6-dione (VII), yield 81.5%, bp 87-88.5° (1 mm), nD 1.4725. The structure of the prepared δ -diketones was confirmed by analysis and by conversion of 3-methylhept-4-ene-2.6-dione into 1.4-dimethylcyclohexan-2-one. Hydropalladium on barium sulfate yield 3-methylhepta-

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Reaction of $oldsymbol{\mathcal{B}}$ -Chlorovinyl Ketones With B -Carbonyl Compounds. XI SOV/79-30-3-42/69 -2.6-dione (VIII), yield 94%, bp 79-80.5° (5 mm), 1.4353, which when treated with 10% solution of sodium hydroxide at 30°, yields 1,4-dimethylcyclohex-1-ene-3-one (IX), yield 70%, bp 76-78° (8 mm), 1.4967. The hydrogenation of the latter over palladium on bariumsulfate yields 1,4-dimethyl-cyclohexan-2-one (X), bp 176-177° (745 mm), n20 $\mathrm{CH_3COCH_{2=}CHCH(CH_3)COCH_3} + \mathrm{H_2} \longrightarrow \mathrm{CH_3COCH_4CH_2CH(CH_3)CCC^{-1}} \longrightarrow$ Card 4/5 There are 18 references, 11 Soviet, 1 U.S., 1 U.K.,

Reaction of Gompounds. XI.

78288 80V/79-30-3-42/69

3 German, 2 French. The U.S. and U.K. references are: Hauser, C., Adams, J., J. Am. Chem. Soc., 66, 345 (1944); Harding, V., Havorth, W., Perkin, W. H., J. Chem. Soc., 93, 1970 (1908).

ASSOCIATION:

Institute of Pharmacology and Chemotherapy of the Academy of Medical Sciences of the USSR (Institut farmakologii i khimioterapii Akademii meditsinskikh nauk SSSR)

SUBMITTED:

March 24, 1959

Card 5/5

GOTTIKH, B. P., BRUSOV, YU. N., KARPEYSKIY, M. YA., KHOMUTOV, R. M., SEVERIN, YE. S. (USSR)

"Synthesis of Certain Biologically Active Hydroxylamine Derivatives."

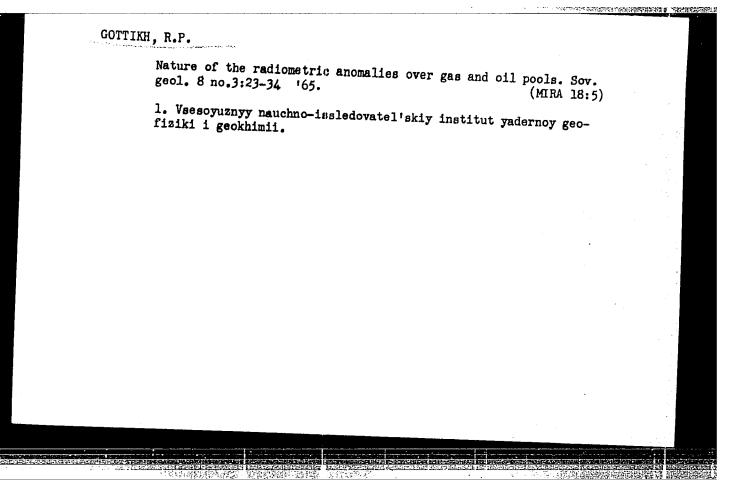
Report presented at the 5th International Biochemistry Congress, Moscow, 10-16 August 1961

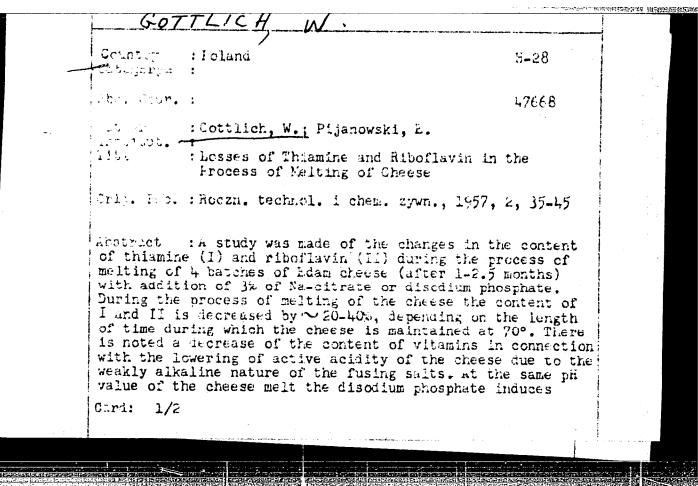
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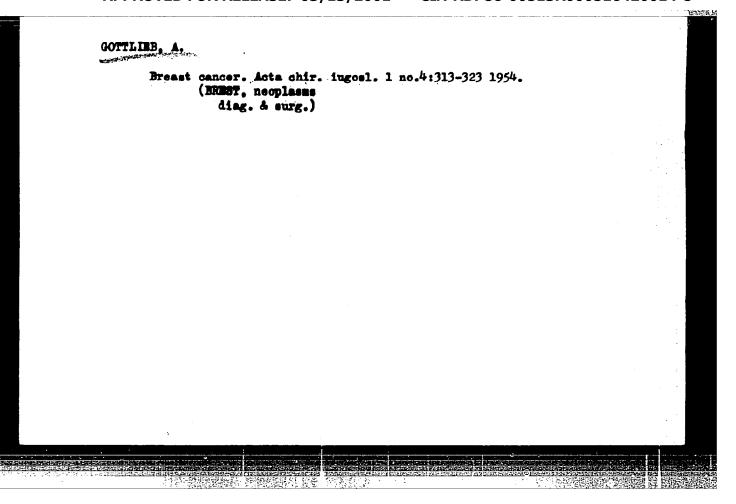
SLUTSKIY, O.I.; GOTTIKH, B.P.

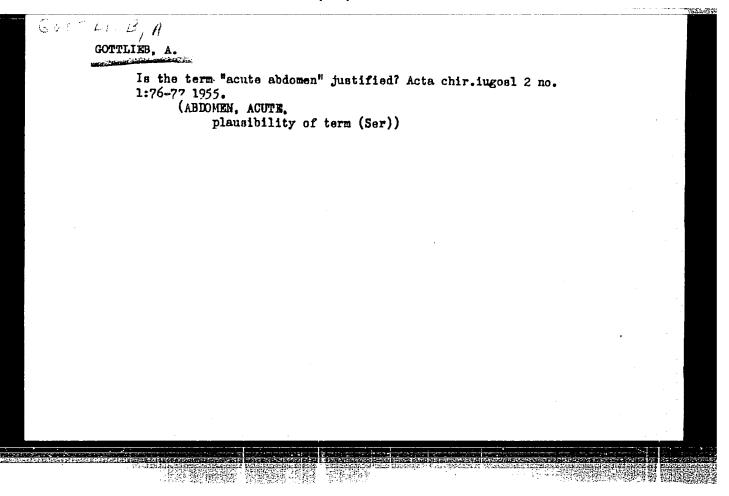
Interaction between N-triphenylmethylglycine and transfer RNA. Biokhimila 30 no.5:1032-1036 S-0 65.

1. Institut radiatsionnoy i fiziko-khimicheskoy biologii AN SSSR,









Syndrome of dissociation of forces between the nucleus pulposus and the system of fibers of the fibrous ring, as a casual factor in intervertebral disk disorders. Med. int., Bucur. 9 no.1:80-89 Jan 57.

(INTERVENTERRAL DISH, diseases etiol., synd. of dissociation of forces between nucleus pulposus & system of fibers of fibrous ring)

RUMANIA

COTTLIBE, E., Biologist and CIOACA, M., Rmg [affiliation not given]

"A New Vegetable Hormone: Giberelline. Prospective Applications in Horticulture."

Bachtrest, Natura, Seria Riclosie, Vol 15, No 1, Jan-Feb 1963, pp 86-88.

Abstract: Reviews the results obtained by the Chemical-Pharmaceutical Research Institute of Bucharcst with regard to the effectiveness of Rumanian giberelline obtained from the following experimental varieties: Felorsonium zonale (4 varieties); P. pelietum and P. Grandlaron; Everancea hortensia (European variety); Usl-ceolaria; Saint Paulia; and Cyclamon. The results given by envier researchers and reported in the literature are confirmed.

Excludes 3 figures and 4 Russian, 1 German and 2

Wastern references.

APRROVED FOR RELEASE: 03/13/2001 CIA-RDR86:00513R000516420014-8

Rumania

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Clinic for Infantile Surgery, Gr. Alexandrescu Mospital, (Clinica de Chirurgie Infantila, Spitalul Gr. Alexandrescu), Bucharest.

Bucharest, Viata Medicala, No 1, Jan 63, pp 59-60.

"Plaster Cast for Facilitating the early Treatment of Congenitally Crooked Legs."

GottLieb, F.

RUMANIA

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Clinic for Infantile Surgery, Gr. Alexandrescu Hospital (Clinica de Chirurgie Infantila, Spitalul Gr. Alexandrescu), Bucharest.

Bucharest, Viata Medicala, No 1, 1 Jan 63, pp 59-60.

"Plaster cast for facilitating the early treatment of congenitally crooked legs."

